

A NEW ANALYSIS OF DUNDASITE FROM TASMANIA

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Dundasite was first described by W. F. Petterd in 1893, and an analysis of Tasmanian material was published in 1910. This analysis shows Fe_2O_3 as a large impurity, with high Al_2O_3 due to probable admixture of gibbsite which is an associated mineral at Dundas. Analyses of relatively uncontaminated material from elsewhere have been presented in the literature (Prior 1906, Palache *et al.* 1951, Beaumont *et Guillemin* 1960) with Beaumont and Guillemin deriving a formula for material from Sardinia of $\text{Pb}_2\text{Al}_2(\text{CO}_3)_4(\text{OH})_8 \cdot 3\text{H}_2\text{O}$ rather than $\text{PbAl}(\text{CO}_3)_2(\text{OH})_4 \cdot 2\text{H}_2\text{O}$ as suggested by Prior. The analysis was performed by the Australian Mineral Development Laboratories on handpicked material taken from a specimen of dundasite (No. 14134) presented to the University of Tasmania with a collection of crocoite from Dundas.

It agrees well with the proposed formula.

	Dundasite	$\text{Pb}_2\text{Al}_2(\text{CO}_3)_4(\text{OH})_8 \cdot 3\text{H}_2\text{O}$
PbO	46.60	46.87
Al_2O_3	21.50	21.41
Fe_2O_3	00.02
CO_2	18.10	18.48
H_2O^+	13.90	13.24
H_2O^-	00.08
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	100.20	100.00

Analyst: C. R. Edmond.

The greatest differences occur between H_2O^+ and CO_2 . As these are complementary it is probable that some $(\text{OH})^-$ replaces $(\text{CO}_3)^{2-}$ in the structure.

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